## Cytotoxicity Studies of the Chemical Constituents from Freshwater Algae *Chara baltica*

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### Tatipamula et al.: Anticancer activity of Chara baltica

For the first time, 89 chemical constituents were identified by gas chromatography-mass spectrometry analysis of ethanol extract of *Clathria baltica*. The chemical examination of ethanol extract of *Clathria baltica* yielded three known metabolites, 1, 2 and 3, for first time from the marine algae *Clathria baltica*. All three metabolites and the ethanol extract of *Clathria baltica* were screened against cancer cell lines, MCF-7, DLD-1, HeLa, FADU, A549, and SKOV3, and one normal human cell line using sulforhodamine B assay and doxorubicin as the standard. Among all isolates, compound 3 showed significant growth inhibition of MCF-7, DLD-1 and FADU with IC<sub>50</sub> values of 26.5, 15.5 and 16.5  $\mu$ g/ml, respectively. Whereas, all the metabolites and the extract exerted lower growth inhibitory effect on the normal human cell line tested. This is the first *in vitro* gas chromatography-mass spectrometry analysis, as well as, cytotoxicity report on marine algae *Clathria baltica*.

Key words: GC-MS analysis, isolation, characterization, anticancer activity, sulforhodamine B assay

Globally, cancer still remains an aggressive killer that severely effecting the human population<sup>[1]</sup>. According to the report of the International Agency for Research on Cancer (IARC), it was documented that within 5 y of diagnosis - 9.6 million cancer deaths from 33.7 million people living with cancer has taken place and 18.1 million new cancer cases were reported in 2018 across the world<sup>[2]</sup>. In addition, the IARC is also estimated that about 26 million new cancer cases and 17 million cancer deaths per year may be recorded by 2030<sup>[3]</sup>. In addition, the present therapies for cancer include chemotherapy and radiotherapy, which cause lot of strain to patients and ultimately damage their health<sup>[4]</sup>. Hence, researchers are focussing on developing new, safer and effective drugs to treat cancer.

From the dawn of ancient medicine, natural products have been used to treat many deadly diseases like cancer. In addition, from the past 30 y, natural products have received great attention in attaining potential as well as novel cancer therapeutic agents<sup>[5,6]</sup>. Moreover, the mechanism of action of inhibiting various stages

of tumorigenesis by most of the natural products were also well-documented<sup>[6]</sup>.

In folklore, marine algae (seaweeds) are used as food, feed, fuel and livelihood in Asian countries. Based on the pigmentation, marine algae are classified as red, brown and green algae. From the past few decades, marine algae have been highly acknowledged to possess noticeable pharmacological activities, which include antineoplastic, antimicrobial and antiviral activities due to their specific functional compounds (which are not available in other plants)<sup>[7-10]</sup>. Particularly, marine algae of *Chara* genus (family: Characeae- charophyte green algae) reported to possess and antioxidant enzyme, allelophatic activities and lipid peroxidation capabilities along with high levels of photosynthetic pigment content<sup>[11]</sup>. Earlier reports from our laboratory showed phytochemical analysis along with antioxidant

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activities of marine algae *Chara baltica*<sup>[12]</sup>. Till date, complete chemical and pharmacological evaluation has not been reported on *C. baltica*. Based on the aforementioned properties of *Chara* genus, as well as, *C. baltica*, *C. baltica* specimens (in July 2018) were subjected to gas chromatography-mass spectrometry (GC-MS) analysis, chemical and cytotoxicity evaluation, which are reported in this research communication.

## **MATERIALS AND METHODS**

The specimens of seaweed *Chara baltica* was collected near Korangi coast, Kakinada, Andhra Pradesh, India at a depth of 8-10 ft (16°80'N and 82°08'E with 3 m elevation) on 3 July, 2018. The seaweed was authenticated in the Botany Department, Andhra University, Visakhapatnam, India. A voucher specimen was deposited in the Marine Organisms Museum in the University College of Pharmaceutical Sciences, Andhra University, Visakhapatnam, India (No. AU-SW-2018-947). All chemicals used in the present study were of analytical grade. Streptozotocin was from Himedia Laboratories Pvt. Ltd. (Mumbai, India); glibenclamide from the Aventis Pharma Ltd. (Mumbai, India); and rat feed from the Hindustan Lever Ltd. (Mumbai, India).

### **Extraction and isolation:**

The seaweed was cleansed from extraneous substance and stored in ethanol-water (9:1) at the site of collection. The crude plant material (100 g) was extracted at least thrice with ethanol-water and concentrated at a reduced pressure and the combined extracts were lyophilized to obtain ethanol extract of C. baltica; as a reddish brown gummy residue (3.25 g). About 2 g of ethanol extract was subjected to column chromatography, afforded three fractions. Fraction I (510 mg) and fraction II (800 mg) were subjected to repeated column chromatography using hexane and ethyl acetate as a solvent system yielded 1 (240 mg) as a colourless liquid and 2 (330 mg) as a pale greenish liquid, respectively. Fraction III (400 mg) were purified by recrystallization technique using hexane and acetone as solvents yielded 3 (200 mg) as a greenish crystals.

Compound 1 (dihydrofuran-2,5-dione), colorless crystals, molecular formula:  $C_4H_4O_3$ ; yield: 240 mg;  $R_{f^*}$  0.7 (hexane:ethyl acetate, 1:9); <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz)  $\delta$  4.00 (4H, s, 2,3-CH<sub>2</sub>); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 400 MHz)  $\delta$  22.16 (C-2/C-3), 170.24 (C-1/C-4); LC-MS *m/z*: 100.

Compound 2 (3-hydroxy-4,4-dimethyldihydrofuran-2(3H)-one), pale greenish liquid,  $C_6H_{10}O_3$ ; yield: 330 mg;  $R_{f^*}$  0.6 (hexane:ethyl acetate, 1:4); <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz)  $\delta$  1.00 (6H, s, 5,6-CH<sub>3</sub>), 2.31 (1H, s, 2-CH), 3.86-3.88 (1H, d, *J*= 8 Hz, 4a-CH<sub>2</sub>), 4.04-4.06 (1H, d, *J*= 8 Hz, 4a-CH<sub>2</sub>); 4.11 (1H, s, 2-OH); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 400 MHz)  $\delta$  23.15 (C-5/C-6), 43.40 (C-3), 80.44 (C-2), 87.23 (C-4), 182.11 (C-1); LC-MS *m/z*: 130.

Compound 3 (2,4-dihydroxy-2,5-dimethylfuran-3(2H)-one), pale greenish crystals, molecular formula:  $C_6H_8O_4$ ; yield: 200 mg;  $R_{f^2}$  0.4 (hexane:ethyl acetate, 1:4); melting point: 163-164°; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz)  $\delta$  1.87 (3H, s, 6-CH<sub>3</sub>), 2.05 (3H, s, 5-CH<sub>3</sub>), 3.62 (1H, s, 3-OH), 3.92 (1H, s, 1-OH); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 400 MHz)  $\delta$  12.14 (C-6), 22.41 (C-5), 99.46 (C-1), 139.32 (C-3), 159.75 (C-4), 192.74 (C-2); CHNS analysis: found C-50.00, H-5.60 (%), calcd. C, 50.46, H, 5.54 (%); LC-MS *m/z*: 144.

# Gas chromatography-mass spectrometry (GC-MS) analysis:

The phytochemical investigation of ethanol extract of C. baltica was performed on a GC-MS system (GCMS-QP2010 Plus, Shimadzu, Europe). Experimental parameters set for the GC-MS were, column oven temperature- 50°; injection temperature- 250°; injection mode- split; flow control mode- linear velocity; pressure- 29.7 kPa; total flow- 7.3 ml/min; column flow- 0.72 ml/min; linear velocity- 30.7 cm/s; purge flow- 3.0 ml/min; split ratio- 5.0; oven temperature program- 50° with rate of increase of 10°/min, 50-300° with rate of increase of  $7.5^{\circ}$ /min, hold time of 2.0 min; equilibrium time- 1.0 min; ion source temperature-210°; interface temperature- 250°; solvent cut time-3.0 min; detector gain- 0.99+0.20 kV; threshold- 1000; start time- 4.0 min; end time- 37.30 min; ACQ modescan; event time- 0.50 s; scan speep- 2500; start m/z-34.0 and end m/z- 1090.00.

## Cytotoxicity assay:

MCF-7 (breast), DLD-1 (colon), HeLa (cervical), FADU (head and neck), A549 (lung), SKOV3 (ovary) and normal human mammary epithelial (NHME) were kindly provided by National Centre for Cell Science, Pune. The cancer cells were maintained in minimum essential medium (MEM), containing 10 % fetal calf serum (FBS), 5 % mixture of 100 units penicillin and 100  $\mu$ g/ml streptomycin in presence of 5 % carbon

dioxide in an incubator with 90 % humidity at  $37^{\circ}$  for 72 h.

Cancer cell lines were maintained in MEM (adjusted to 10 % FBS, 1.5 g/ml NaHCO<sub>2</sub>, 0.1 mM MEM nonessential amino acids and 1 mM sodium pyruvate). Three days prior to performing assay, the cells were washed with sterilized phosphate-buffered saline and grown using MEM media (supplemented with 0.25 % trypsin in versene-EDTA and 10 % FBS) and mixed to obtain homogeneous suspension of cells. The suspension was taken in a sterilized polypropylene tube and the cell count in each well was determined in a hemacytometer chamber under a microscope using 0.4 % trypan blue solution. The minimal seed density must be  $1.9 \times 10^4$  cells per well. All crude extracts and the standard were dissolved in dimethyl sulfoxide (DMSO) to 100 mg/ml and 10 µg/ml, respectively. Doxorubicin and DMSO were used as a standard and control, respectively.

The sulforhodamine B (SRB) assay<sup>[13]</sup> is based on the estimation of cellular protein content. The prepared samples were taken in 96-well tissue-culture plate and added 190 µl screened ideal cell suspension and mixed occasionally and incubate at 37° with 5 % CO<sub>2</sub> and 90 % relative humidity for 3 h. Then 100 µl cold TCA was added to each well and incubated at 4° for 1 h. After that the plates were gently washed using water, dried using blow dryer and air-dried at room temperature. To each completely dried well, 100 µl of 0.057 % SRB solution was added, kept aside for 30 min and quickly rinse with 1 % acetic acid. To the dried plate, 200 µl of 10 mM Tris base (pH 10.5) solution was added, shake for 5 min and measure the OD at 510 nm. The blank contains only medium while the control has only cancer cells with no test samples. Percent growth inhibition was calculated using the formula, growth inhibition  $(\%) = 100 - [(S-B)/(C-B)] \times 100$ , where S is mean OD value of sample, B is mean OD value of blank, C is mean OD value of control.

## **RESULTS AND DISCUSSION**

After the successful extraction of the whole marine algae material with ethanol-water, the dried ethanol extract was subjected to GC-MS analysis. The GC-MS chromatogram revealed the presence of various chemical components with different retention times, whereas the MS analyzes the compounds eluted at different times to detect its nature and structure of the compounds. The results pertaining to GC-MS analysis of the ethanol extract lead to the identification of a number of compounds. A total of 86 compounds were identified by GC-MS analysis of ethanol extract, which were tabulated in Table 1.

The collected seaweed was extracted with ethanolwater at room temperature. The obtained ethanol extract *C. baltica* was subjected to chromatographic purification led to isolation of three metabolites (1-3, fig. 1). Compound 1 was obtained as a colorless crystals with  $R_f$  value 0.7 (hexane:ethyl acetate, 1:9) and the LC-MS ion peak at m/z 100 confirmed the molecular formula as  $C_4H_4O_3$ . The <sup>1</sup>H and <sup>13</sup>C NMR data showed the presence of two methane groups at  $\delta_H$  4.00 (*s*, 4H) with corresponding carbon signal at  $\delta_C$ 22.16 (C-2 and C-3). In addition, the <sup>13</sup>C NMR also revealed the presence of dione groups at  $\delta_C$  170.24 (C-1 and C-4). Hence, compound 1 was confirmed after corroboration with the existing literature<sup>[14]</sup> as a dihydrofuran-2,5-dione (fig. 1).

Compound 2 was obtained as a pale greenish liquid. Based on the LC-MS ion peak at m/z 130 confirmed the molecular formula as  $C_6H_{10}O_3$ . The <sup>1</sup>H NMR showed the presence of two singlets for two methyl groups at  $\delta_H$  1.00 (*s*, 6H) with corresponding carbon signal at  $\delta_C$ 23.15 (C-5 and C-6), including one methine group at  $\delta_H$  2.31 (*s*, 1H) with corresponding carbon signal at  $\delta_C$ 80.44 (C-2), and one hydroxyl group at  $\delta_H$  4.11 (*s*, 1H), additionally, two doublets at  $\delta_H$  3.86-3.88 (*d*, *J*=8 Hz, 1H) and 4.04-4.06 (*d*, *J*=8 Hz, 1H), which confirmed the presence of methane group at  $\delta_C$  87.23. In addition, the <sup>13</sup>C NMR also revealed the presence of dione group at  $\delta_C$  182.11 (C-1). By corroboration with the existing literature compound 2 was found to be 3-hydroxy-4,4dimethyldihydrofuran-2(3H)-one (fig. 1)<sup>[15]</sup>.

Compound 3 was obtained as a pale greenish crystals, based on the LC-MS ion peak at m/z 144 confirmed the molecular formula as  $C_6H_8O_4$ . The <sup>1</sup>H NMR showed the presence of four singlets for two methyl groups at  $\delta_H$  1.87 (*s*, 3H) and 2.05 (*s*, 3H), and for two hydroxyl groups at  $\delta_H$  3.62 (*s*, 1H) and 3.92 (*s*, 1H). The <sup>13</sup>C NMR revealed the presence of two methyl group at  $\delta_C$ 12.14 (C-6) and 22.41 (C-5), two hydroxyl groups at  $\delta_C$ 99.46 (C-1) and 139.32 (C-3), dione group at  $\delta_C$  192.74 (C-2), and one methene carbon signal at  $\delta_C$  159.75 (C-4). Based on the aforementioned data, compound 3 was compared with the existing literature on lichen metabolites and it was confirmed as a 2,4-dihydroxy-2,5-dimethylfuran-3(2H)-one (fig. 1). Compound 3 was earlier reported from *Sysepalum dulcificum*<sup>[16]</sup>.

# TABLE 1: CHEMICAL CONSTITUTENTS IDENTIFIED BY GC-MS ANALYSIS OF ETHANOL EXTRACTS OF CHARA BALTICA

S. No	Compound name	Chemical structure	Molecular formula
1	(2Z)-1,3-dimethyl-6-oxo-2-hexenyl acetate		$C_{10}H_{16}O_{3}$
2	(2Z)-5-methyl-2-decene		C <sub>11</sub> H <sub>12</sub>
3	(9E)-9-octadecenal		C <sub>18</sub> H <sub>34</sub> O
4	(9Z)-9-octadecenamide		C <sub>18</sub> H <sub>35</sub> NO
5	1-((E)-[(2-methylphenyl)imino] methyl)-2-naphthol	С С С С С С С С С С С Н	C <sub>18</sub> H <sub>15</sub> NO
6	1,2,5-trimethyl-4-piperidinone thiosemicarbazone	$\underbrace{\begin{array}{c} & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & &$	$C_9H_{18}N_4S$
7	1,3-cyclohexanedione		$C_6H_8O_2$
8	1,4-dimethyl-5-oxabicyclo[2.1.0] pentane		C <sub>6</sub> H <sub>10</sub> O
9	1,5-anhydro-6-deoxyhexo-2,3- diulose	ОН	C <sub>7</sub> H <sub>10</sub> O <sub>3</sub>
10	10-undecen-1-ylester		C <sub>17</sub> H <sub>32</sub> O <sub>2</sub>
11	1-chloro-4-decyne	ci~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	C <sub>10</sub> H <sub>17</sub> Cl
12	1-hexadecylpyridinium chloride monohydrate		C <sub>21</sub> H <sub>40</sub> ClNO
13	1-Tetradecanol, acetate		C <sub>16</sub> H <sub>32</sub> O <sub>2</sub>
14	1-undecene		C <sub>11</sub> H <sub>22</sub>
15	3-hydroxy-4,4- dimethyldihydrofuran-2(3H)-one		C <sub>6</sub> H <sub>10</sub> O <sub>3</sub>
16	2-(benzyloxy)-4-bromo-1,3- butanediol	OH OH OH	C <sub>11</sub> H <sub>15</sub> BrO <sub>3</sub>
17	2,10-dimethylundecane		C <sub>13</sub> H <sub>28</sub>
18	2,2,4-trimethyl-1-pentanol	он	C <sub>8</sub> H <sub>18</sub> O
19	2,2-dimethyl-1-phenyl-1-propanol	OH	C <sub>11</sub> H <sub>16</sub> O

20	2,2-dioctyl-1-oxohydrazine-1-oxide	0 <sub>N+</sub> 0	$C_{16}H_{34}H_2O_2$
21	9,12-Octadecadienoic acid (Z,Z)-, 2,3-bis(acetyloxy)propyl ester	$\begin{array}{c} 0 \\ \downarrow 0 \\ \downarrow$	$C_{25}H_{42}O_{6}$
22	2,3-dihexyl-2-cyclopropene-1- carboxylic acid	O OH	C <sub>16</sub> H <sub>28</sub> O <sub>2</sub>
23	2,4,6-trimethyldecane		$C_{13}H_{28}$
24	2,4-Dihydroxy-2,5-dimethyl-3(2H)- furanone		$C_6H_8O_4$
25	2,5-furandione		$C_4H_2O_3$
26	2-acetoxytetradecane		$C_{16}H_{32}O_{2}$
27	2-acetoxytridecane		C <sub>15</sub> H <sub>30</sub> O <sub>2</sub>
28	2-cyclohexen-1-one		C <sub>6</sub> H <sub>8</sub> O
29	2-cyclohexylethyl acetate		C <sub>10</sub> H <sub>18</sub> O <sub>2</sub>
30	2-hydroxy-3-methylbutanoic acid		$C_{5}H_{10}O_{3}$
31	2-propylheptan-1-ol	ОН	C <sub>10</sub> H <sub>22</sub> O
32	3-(methoxymethoxy)-butanoic acid	~0~0↓↓0H	C <sub>6</sub> H <sub>12</sub> O <sub>4</sub>
33	3,5,5-trimethyl-4-(3-oxo-1- butenyl)-2-cyclohexen-1-one		C <sub>13</sub> H <sub>18</sub> O <sub>2</sub>
34	3,5-bis(1,1-dimethylethyl)-phenol	Ч СН ОН	C <sub>14</sub> H <sub>22</sub> O
35	3,6-dimethyl-2-octanone		C <sub>10</sub> H <sub>20</sub> O
36	3-acetoxydodecane		$C_{14}H_{28}O_{2}$
37	3-ethyl-2,5-dimethylhexane		C <sub>10</sub> H <sub>22</sub>
38	3-hexene-2,5-diol	ОН	C <sub>6</sub> H <sub>12</sub> O <sub>2</sub>
39	3-hydroxy-2,2,6- trimethylcyclohexyl-methyl acetate	о с с с с с с с с с с с с с с с с с с с	$C_{12}H_{22}O_{3}$

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40	3-hydroxydodecanoic acid	он о	C <sub>12</sub> H <sub>24</sub> O <sub>3</sub>
41	3-nonyn-1-ol	ОН	C <sub>9</sub> H <sub>16</sub> O
42	4-acetoxypentadecane		C <sub>17</sub> H <sub>34</sub> O <sub>2</sub>
43	4-heptanol	ОН	C <sub>7</sub> H <sub>16</sub> O
44	5-(1-benzyl-1h-indol-3- ylmethylene)-1-(2-ethyl-phenyl)- pyrimidine-2,4,6-trione		$C_{28}H_{23}N_3O_3$
45	5-bromo-3-methylpentyl acetate	0 L O Br	C <sub>8</sub> H <sub>15</sub> BrO <sub>2</sub>
46	5-cyclopropylidene-1-pentanol	ОН	$C_8H_{14}O$
47	7,9-ditert-butyl-1-oxaspiro[4.5] deca-6,9-diene-2,8-dione		C <sub>17</sub> H <sub>24</sub> O <sub>3</sub>
48	9-octadecenamide		C <sub>18</sub> H <sub>35</sub> NO
49	9-octadecenoic acid	ОН	$C_{18}H_{34}O_{2}$
50	Acetic acid	ОЦОН	$C_2H_4O_2$
52	Araldite		C <sub>16</sub> H <sub>22</sub> O <sub>4</sub>
53	Benzenemethanol	ОН	C <sub>7</sub> H <sub>8</sub> O
54	Benzoic acid	ОН	C <sub>7</sub> H <sub>6</sub> O <sub>2</sub>
55	Butanoic acid	о	$C_4H_8O_2$
56	Cis-oleic acid	О	C <sub>18</sub> H <sub>34</sub> O <sub>2</sub>
57	Cyclopentaneundecanoic acid	О	C <sub>16</sub> H <sub>30</sub> O <sub>2</sub>
58	Dimethyl(ethenyl)propoxysilane	si on	C <sub>7</sub> H <sub>16</sub> OSi
59	Di-n-butyl sulphite		C <sub>8</sub> H <sub>18</sub> O <sub>3</sub> S
60	Di-n-octyl phthalate		C <sub>24</sub> H <sub>38</sub> O <sub>4</sub>

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61	DL-alanine	о он	C <sub>3</sub> H <sub>7</sub> NO <sub>2</sub>
62	E-2-undecen-1-ol	NH <sub>2</sub> OH	C <sub>11</sub> H <sub>22</sub> O
63	Ethyl-(4E)-2-acetyl-2-methyl-4- hexenoate		C <sub>11</sub> H <sub>18</sub> O <sub>3</sub>
64	Ethyl docosanoate		$C_{24}H_{48}O_{2}$
65	Ethyl nonadecanoate		$C_{21}H_{42}O_{2}$
66	Hexadecyl acetate		C <sub>18</sub> H <sub>36</sub> O <sub>2</sub>
67	Methyl-(E)-docos-13-enoate		$C_{23}H_{44}O_{2}$
68	Methyl-(9E)-9-octadecenoate		$C_{19}H_{36}O_{2}$
69	Methyl-3-(3,5-di-tert-butyl-4- hydroxyphenyl)propionate		C <sub>18</sub> H <sub>28</sub> O <sub>3</sub>
70	Methyl-3-isopropyl-6-oxoheptanoate		C <sub>11</sub> H <sub>20</sub> O <sub>3</sub>
71	Methyl-4-hydroxy-3-methyl-2- butenoate		C <sub>6</sub> H <sub>10</sub> O <sub>3</sub>
72	Methyl heptanoate		$C_8 H_{16} O_2$
73	n-cyclohexyl methacrylamide		C <sub>10</sub> H <sub>17</sub> NO
74	n-decyl acetate		C <sub>12</sub> H <sub>24</sub> O <sub>2</sub>
75	n-heptadecane	~~~~~	$C_{17}H_{36}$
76	Octadecane	$\sim\!\!\!\sim\!\!\!\sim\!\!\!\sim\!\!\!\sim\!\!\!\sim\!\!\!\sim\!\!\!\sim\!\!\!\sim\!\!\!\sim\!$	C <sub>18</sub> H <sub>38</sub>
77	Octadecanoic acid	О	C <sub>18</sub> H <sub>38</sub> O <sub>2</sub>
78	Octanal dimethyl acetal		C <sub>10</sub> H <sub>22</sub> O <sub>2</sub>
79	Palmitic acid	О	C <sub>16</sub> H <sub>32</sub> O <sub>2</sub>
80	Palmitic acid ethyl ester		$C_{18}H_{36}O_{2}$
81	Pelargonic acid methyl ester		$C_{10}H_{20}O_{2}$
82	Pentanoic acid	ОН	$C_5 H_{10} O_2$
83	Phthalic acid		$C_8H_6O_4$
84	Tert-nonanethiol	SH	$C_9H_{20}S$



#### TABLE 2: CYTOTOXICITY STUDIES OF COMPOUNDS 1-3 AND ETHANOL EXTRACT OF CHARA BALTICA

Sample	Percent growth inhibition at 30 µg/ml							
	MCF-7	DLD-1	HeLa	FADU	A549	SKOV3	NHME <sup>c</sup>	
1	1.46±0.18	5.64±0.65	3.97±0.70	13.80±1.06	21.37±1.94	5.69±0.87	0.52±0.01	
2	7.17±0.24	11.91±0.50	5.12±0.18	16.41±1.26	14.57±0.39	8.14±1.05	1.04±0.16	
3	56.86±4.78	76.37±5.34	22.27±6.23	64.08±3.92	34.07±4.84	46.11±4.74	9.40±0.33	
Ethanol extract*	71.53±4.71	87.11±3.58	40.90±2.57	70.46±4.96	57.62±5.58	48.24±4.44	2.91±0.31	
Doxorubicin#	84.40±0.80	66.71±0.71	90.71±0.13	88.34±0.27	65.40±0.60	77.05±0.22	10.08±0.95	

Mean±SEM values (n=3); #10 µg/ml; \*100 µg/ml; <sup>c</sup>normal human mammary epithelialium

## TABLE 3: GROWTH INHIBITION AND IC<sub>50</sub> VALUES OF COMPOUND 3 AND ETHANOL EXTRACTS OF CHARA BALTICA

Coll line comple	Concentration (µg/ml)					
Cell line sample —	5	10	20	30	(µ̃g/ml)	
MCF-7						
Compound 3	18.09±2.87	28.17±2.87	38.59±1.74	56.86±4.78	26.5	
Ethanol extracts**	25.06±1.78	36.58±2.68	50.05±3.57	71.53±4.71	75	
Doxorubicin*	30.87±0.61	45.45±0.18	67.08±0.11	81.25±0.27	5.5	
DLD-1						
Compound 3	22.55±4.84	39.77±2.83	57.92±7.78	76.37±5.34	15.5	
Ethanol extract **	29.75±1.77	42.57±2.49	59.22±1.77	87.11±3.58	61	
Doxorubicin*	19.75±4.61	31.58±6.85	44.75±2.84	52.57±7.97	5.4	
FADU						
Compound 3	29.68±2.68	40.88±4.06	54.87±4.54	64.08±3.92	16.5	
Ethanol extract **	26.47±2.97	37.18±2.52	50.17±3.58	70.46±4.96	74.5	
Doxorubicin*	41.01±3.85	52.14±2.84	68.88±1.77	85.55±6.24	4.5	
A549						
Ethanol extract **	26.47±2.97	37.18±2.52	50.17±3.58	57.62±5.58	74.5	
Doxorubicin*	41.01±3.85	52.14±2.84	68.88±1.77	85.55±6.24	4.5	

Each value is a mean+SD of % growth inhibition of the cell line tested with n=3; \*2.5, 5.0, 7.5 and 10 µg/ml concentrations; \*\*25, 50, 75 and 100 µg/ml concentrations



Fig. 1: Known secondary metabolites from ethanol extracts of *Chara baltica* 

Earlier reports of *C. baltica* indicated free radical scavenging activity<sup>[13]</sup>. Based on the report, the isolated metabolites (1-3) were screened along with ethanol extract against six different cancer cell lines using SRB assay with doxorubicin as standard. Initially, compounds (1-3) at 30  $\mu$ g/ml concentration, extract at 100  $\mu$ g/ml concentration and doxorubicin at 10  $\mu$ g/ml

concentration were screened against the cell lines tested and percent inhibition of cell growth was tabulated in Table 2.

Further, the samples that caused 50 % or more of cell death were further screened at 5, 10, 20 and 30  $\mu$ g/ml concentrations for pure compounds; 25, 50, 75 and 100  $\mu$ g/ml concentrations for extracts; 2.5, 5.0, 7.5 and 10  $\mu$ g/ml concentrations for doxorubicin. The results were plotted to obtain IC<sub>50</sub> values. The lower IC<sub>50</sub> value indicated better inhibitory profile against cancer cell lines.

During the initial screening, the isolate 3 and ethanol extract depicted prominent degree of specificity against MCF-7, DLD-1, FADU and A549. Besides, all the samples exhibited very low toxicity towards NHME cell

lines (Table 2). From the outcomes of the SRB assay, it is interesting to note that compound 3 (30  $\mu$ g/ml) and ethanol extract (100  $\mu$ g/ml) showed prominent degree of specificity against DLD-1 with 76.37±5.34 and 87.11±3.58 % growth inhibition which was found to be better than activity of 66.71±0.71 % shown by the standard at 10  $\mu$ g/ml (Table 2).

From the Table 3, the IC<sub>50</sub> values of 3 and ethanol extract against MCF-7 were 26.5 and 75.0  $\mu$ g/ml, respectively, whereas standard value was 5.5  $\mu$ g/ml. In addition, the concentration of 3 and ethanol extract needed for 50 % inhibition of DLD-1 were found to be 15.5 and 61.0  $\mu$ g/ml, respectively, while for doxorubicin it was 5.4  $\mu$ g/ml (Table 3).

From the SRB assay of FADU (Table 3) it can be calculated that the  $IC_{50}$  values of 3 and ethanol extract against FADU were 16.5 and 74.5 µg/ml, respectively, whereas standard value was 4.5 µg/ml. Additionally, the concentration of ethanol extract needed for 50 % inhibition of A549 was found to be 74.5 µg/ml, while standard (doxorubicin) was 4.5 µg/ml (Table 3).

Hence, from the overall SRB assay it can be concluded that the key agent responsible for anticancer activity is compound 3. In addition, the growth inhibitory properties of compound 3 is mainly due to the higher levels of oxygenated substituents present in its chemical structure, this higher levels of oxygenated content help in irradiating free radicals that are usually formed during massive cell division of cancer cells, which ultimately lead to the cell death<sup>[13,17]</sup>.

In conclusion, this is the first report of GC-MS analysis, chemical and cytotoxicity profile of marine algae *C. baltica*. The GC-MS analysis of ethanol extract *C. baltica* revealed presence of 89 chemical constituents. In addition, the chemical examination of ethanol extract showed the presence of three metabolites 1-3. Additionally, the cytotoxicity studies of ethanol extract indicated specificity towards MCF-7, DLD-1, FADU and A549 without affecting NHME. The SRB assay screening proved that compound 3 has greater cytotoxicity. The results of the current study could be useful for further research on cancer to identify potential bioactive molecules from different aquatic fauna, such as marine algae like *C. baltica*.

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### **Conflict of interest:**

No conflict of interest between any of the authors.

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